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## **COLORIMETRIC ESTIMATION OF SUMATRIPTAN SUCCINATE IN TABLET AND ORAL FILMS DOSAGE FORMS USING EHRlich REAGENT**

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### **ABSTRACT**

A simple, sensitive and cost effective visible spectrophotometric method has been developed for the determination of sumatriptan succinate from bulk, tablet dosage forms and oral films. The method is based on the formation of purple colored chromogen by the drug with Ehrlich reagent with an absorption maximum of 571 nm. The Beer's law was obeyed in the concentration range of 20 –100 µg/mL. The proposed method was successfully applied for the estimation of sumatriptan succinate in commercially available tablets and the results were statistically compared with those obtained by the reference method and validated by recovery studies and the percent recovery was found to be more than 99.5%.

## 1. INTRODUCTION

Sumatriptan succinate (SUMA) is the most frequently prescribed anti-migraine drug of triptan class. It is chemically known as 3-[2-(Dimethylamino) ethyl] -N-methyl-1H indole -5-methane sulphonamide succinate (1:1) base [1]. SUM is a specific and selective 5- hydroxyl tryptamine receptor (5- HT1D) agonist with no effect on the other 5HT receptor (5HT2-5 HT7) sub types. It is used widely for prophylaxis and acute relief of migraine attack with or without aura. SUM is official in European Pharmacopoeia [2] and United States Pharmacopoeia [3], which suggests chromatographic methods for the determination of SUM in bulk, tablet formulations and oral films. Several analytical techniques like HPLC [4-9], HPLC-MS-MS [10-13], HPLC- ECD [14- 15], HPLC-coulometer [16], capillary LC-MS-MS [17], HPTLC [18], spectrophotometric with HPTLC [19], RPHPLC with colorimetric [20], UV [21] and voltametry [22], capillary electrophoresis [23], densitometry and spectrophotometric detection [24] have been reported in the literature. However analytical important functional groups in SUMA have not been exploited properly in developing visible spectrometric methods. So the authors have made some attempts in this direction and succeeded in developing a method based on the reaction between the drug and Ehrlich reagent (p-Dimethyl Amino Benzaldehyde) [25].

In this method purple colored species (quinoneimines) was formed rapidly when indoles with hydrogen  $\alpha$  and  $\beta$  to the ring nitrogen react with Ehrlich reagent. The method can be extended for the routine assay of SUMA formulations.

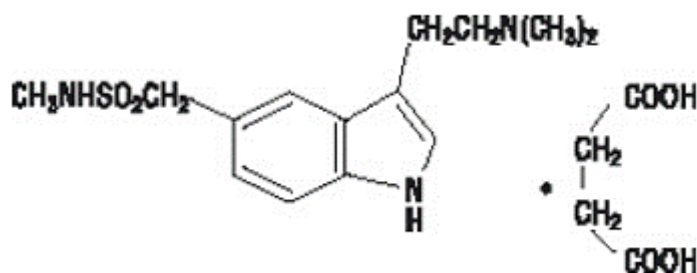


Fig:1 Sumatriptan succinate

## 2. MATERIALS AND METHODS

**2.1 Instruments:** Optical density measurements were made on Systronics UV spectrophotometer 2203.

**2.2 Preparation of reagent solutions:** All the reagents used were of analytical reagent grade.

- a) Ehrlich reagent ( 1%): 1gm of p-Dimethyl Amino Benzaldehydewas dissolved in 100ml of 2.5 N HCl
- b) 2.5 N HCl: 21.25 mL of Conc. HCl is dissolved in 100 mL of distilled water
- c) 1N HCl: 85 ml of Conc. HCl was dissolved in 1000ml of distilled water
- d) Sulphuric acid
- e) Sumatriptan succinate (API) was procured from NATCO labs limited, Hyderabad, Telangana.

**Preparation of standard stock:** The standard stock solution (1 mg/mL) of SUMA was prepared by dissolving 100 mg of SUMA in 100 mL distilled water. The working standard solution of SUMA (100 $\mu$ g/mL) was obtained by appropriately diluting the standard stock solution by using the same solvent a series of standards were freshly prepared during the analysis day.

**Preparation of Oral films of Sumatriptan succinate:**

Initially placebo MDFs were prepared with different polymers like HPMC (E5, E15, and K4M), HPC, MC, Na CMC, PVP, Gelatin, and sodium alginate. Finally, from these trials made and results obtained, HPMC E5 and HPMC E15were selected for further development. In the initial trials 100mg of drug was added to the formulation and the films were prepared. However, crystallization of the drug was observed over a period of time and hence, the drug amounts were adjusted to 50mg per batch. Totally a 5g batch size of formulation gave approximately 100cm<sup>2</sup> film area. Different homogenous films of SUMA were prepared; all the films are transparent, colorless, and soft with no spots found on them.

**Determination of wavelength maximum ( $\lambda_{max}$ ):** The 0.5 mL of working standard solution (100 $\mu$ g/mL) was taken in 5 mL standard flask. To this, 1.0 mL of Ehrlichreagent (1%), 1.0 mL of H<sub>2</sub>SO<sub>4</sub>, and kept aside for 10 min for complete color development. Then the volume was made up to 5mL with 1N HClto get a concentration of 50 $\mu$ g/mL. In order to investigate the wavelength maximum, the above standard solution was scanned in the range of 360–670 nm by UV–Visible spectrophotometer. From the UV spectra (Fig.6), it was concluded that 571.0 nm is the most appropriate wavelength for analyzing SUMA with suitable sensitivity.

**Preparation of calibration curve:** Aliquots of working standard solution (100  $\mu$ g/mL) such as 0.1, 0.2, 0.3, 0.4 and 0.5 mL were taken separately in a series of 5 mL standard flask, to get a concentration of 10, 20, 30, 40 and 50  $\mu$ g/mL, respectively. 1.0 mL of Ehrlich reagent (1%), 1.0 mL of H<sub>2</sub>SO<sub>4</sub>, and kept aside for 10 min for complete color development. Then the

volume was made up to 5mL with 1N HCl to get a concentration range of 10- 50  $\mu\text{g/mL}$ . The absorbance was measured at 571.0 nm against the reagent blank within stability period of 30 min. The calibration graph was constructed by plotting the drug concentration versus absorbance (Fig.3).

**Assay of Sumatriptan succinate in tablet formulation:** About 20 tablets were pulverized and the powder equivalent to 100 mg of SUMA was weighed, dispersed in 25 mL of water, sonicated for 30 min and the volume was made up to 100 mL with distilled water to get a concentration of 1 mg/mL and filtered through Whatmann filter paper No 41. It was used as stock sample solution and was further diluted with the same solvent to get working standard solution.

#### **Assay of Sumatriptan Succinate in oral film formulation**

A film equivalent to 8mg of Sumatriptan succinate was taken and dissolved in 10ml standard flask. The drug was extracted with 5 ml of water and diluted up to mark with the same solvent. Then it was filtered. Working sample is prepared. To 0.5ml of the working sample solution, 1.0 mL of Ehrlich reagent (1%), 1.0 mL of  $\text{H}_2\text{SO}_4$ , and kept aside for 10 min for complete colour development. Then the volume was made up to 5mL with 1N HCl. The absorbance was measured at 571.0 nm against the reagent blank within stability period of 30 min.

### **3. RESULTS AND DISCUSSION**

In the present investigation, the indoles with a hydrogen  $\alpha$  and  $\beta$  to the ring nitrogen react with Ehrlich reagent. The formation of colored species with this reagent may be assigned through above analogy as shown in Fig.2. In developing this method, a systematic study of the effects of various parameters was undertaken by varying the volume of Ehrlich reagent and  $\text{H}_2\text{SO}_4$  and time. The effect of various parameters such as time, volume and strength of Ehrlich reagent of the colored species were studied and the optimum conditions were established and the  $\lambda_{\text{max}}$  was obtained at 571nm and the over lab mode was shown in fig 7.

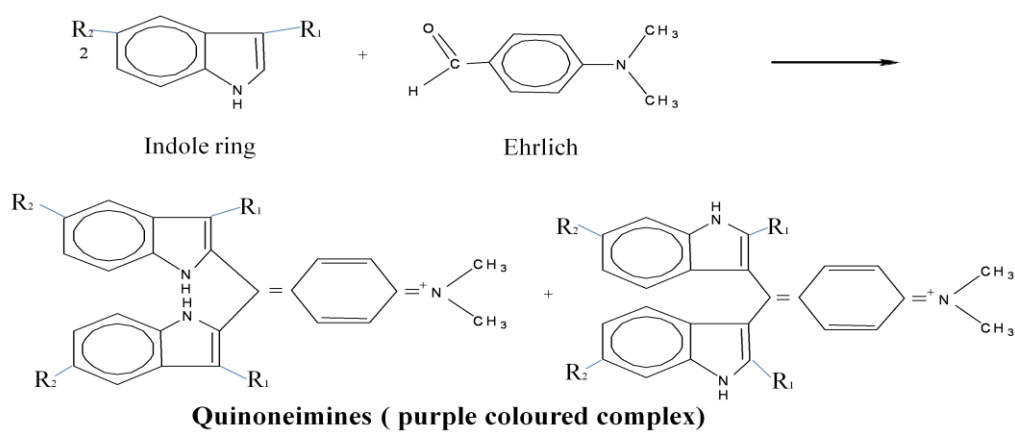
Regression characteristics like standard deviation of slope (Sb), standard deviation of intercept (Sa), were calculated and are shown in Table 1. Commercial formulations containing SUMA were successfully analyzed by the proposed method. The values and the spectra obtained by the proposed method are shown in table 1 and fig4-6. As an additional demonstration of accuracy, recovery experiments were performed by adding a fixed amount of the drug to the preanalyzed formulations at three different concentration levels (80%,

100% and 120%) These results are summarized in Table 2. The ingredients usually present in formulations of SUMA did not interfere with the proposed analytical method.

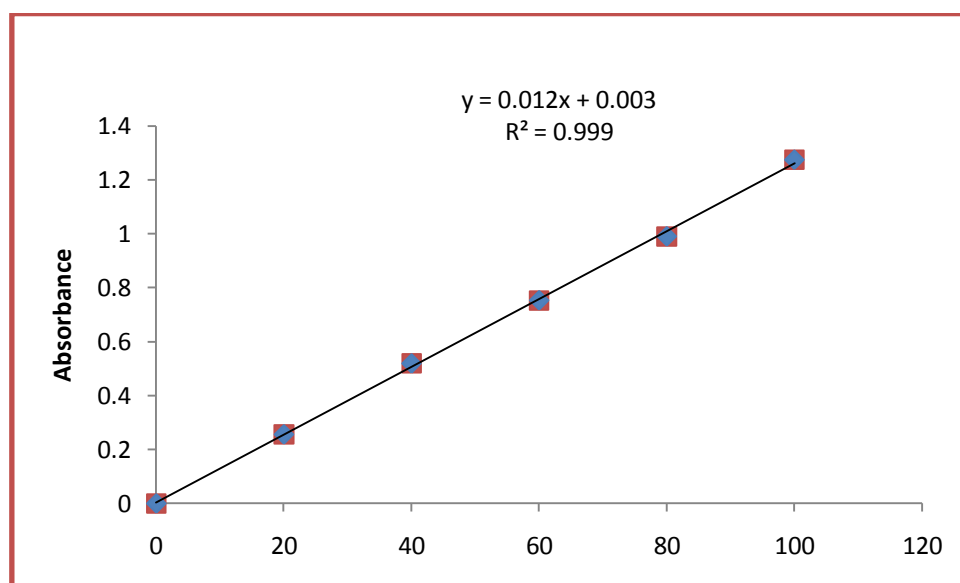
#### 4. CONCLUSIONS

In conclusion, the proposed visible spectrophotometric methods for the estimation of SUM are possess reasonable precision, accuracy, simple, sensitive, and can be used as an alternative method to the reported ones for the routine determination of SUMA.

**Fig :2 Formation of coloured species**



Where  $R_1 = CH_2CH_2N(CH_3)_2$  and  $R_2 = CH_3NHSO_2CH_2$



**Fig:3 Calibration curve of Sumatriptan succinate**

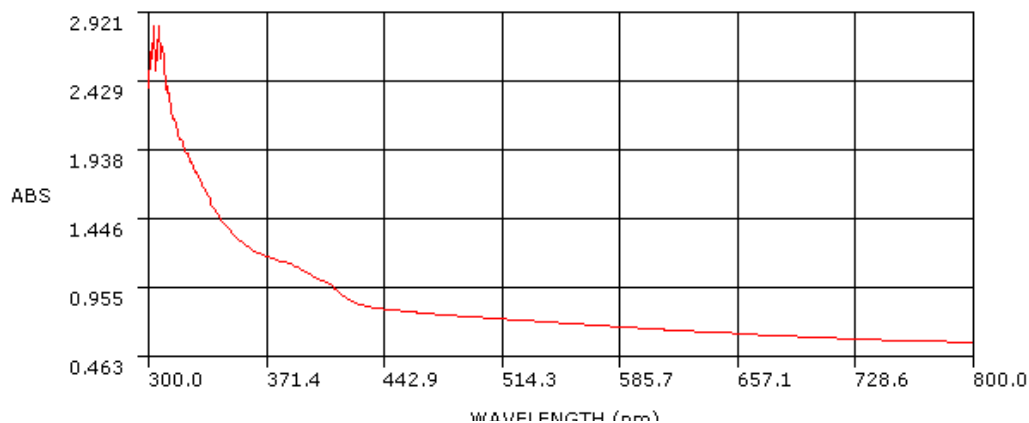


Fig:4 Absorbance spectra of Blank

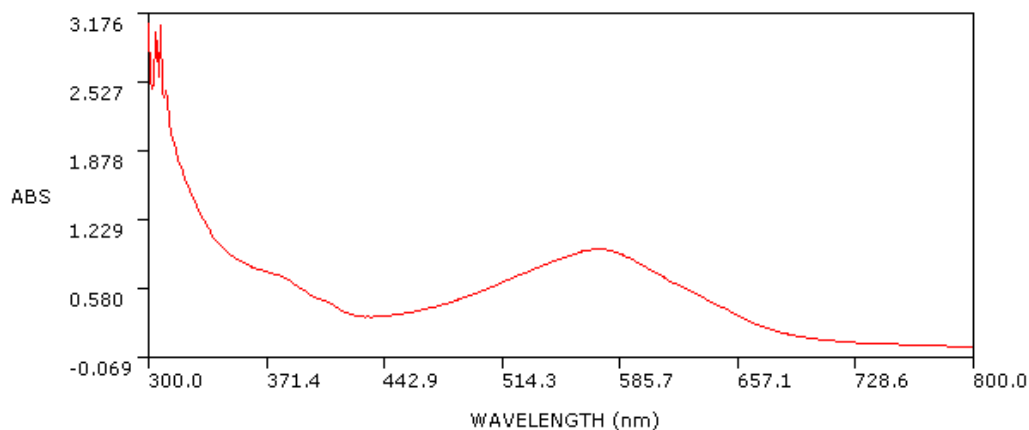


Fig:5 Absorbance Spectra of Sumatriptan succinate in oral film

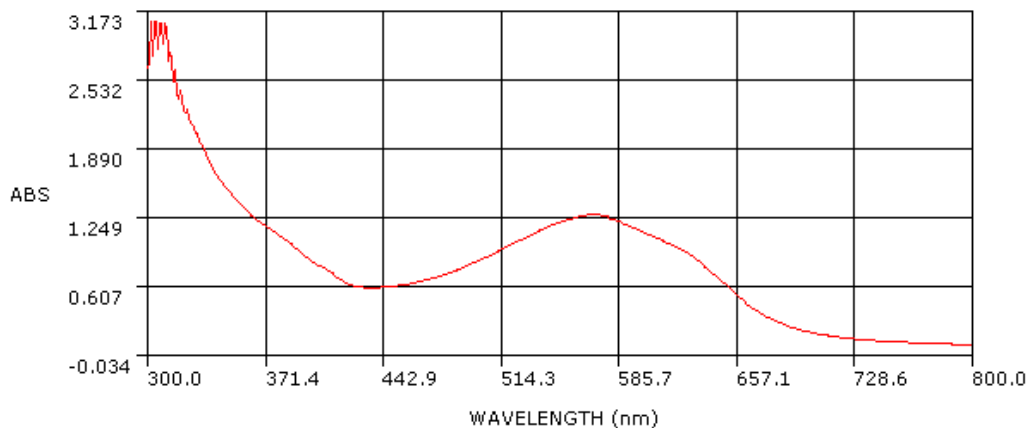


Fig:6 Absorbance spectra of Sumatriptan succinate in tablet dosage form

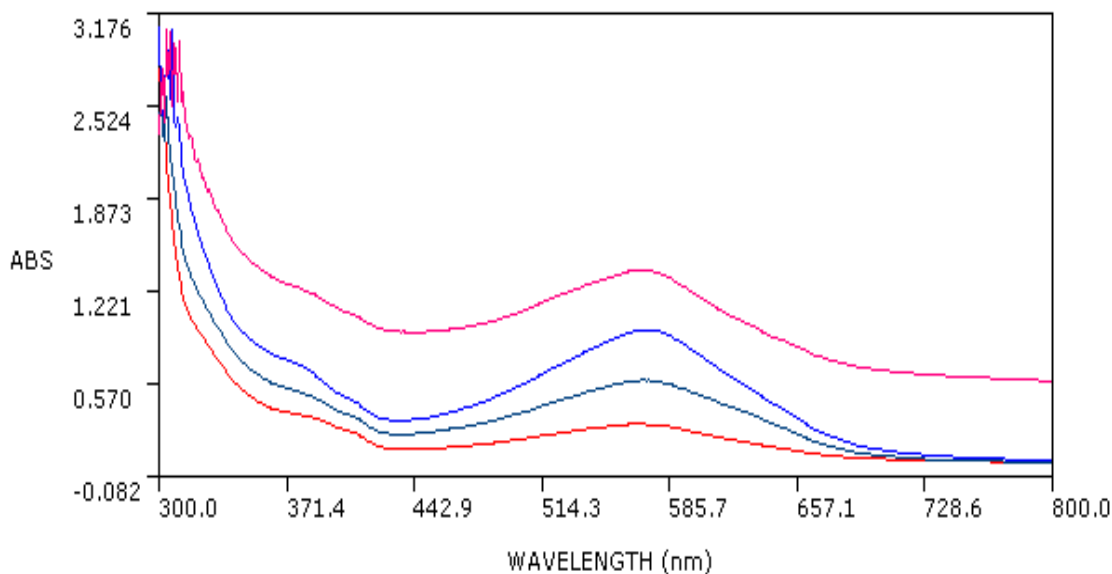


Fig:7 Over lay mode of absorbance spectra

**Table 1: Optical characteristics, precision and accuracy of the proposed method**

Parameters	Sumatriptan succinate Drug
$\lambda$ max	571nm
Range ( $\mu\text{g/ml}$ )	20-100 $\mu\text{g/ml}$
Regression equation ( $y=mx+ c$ )	$Y=0.012x+0.003$
A(1%,1cm)	215
Correlation coefficient ( $r^2$ )	0.999
LOD ( mg/ml)	0.362
LOQ ( mg/ml)	1.099
Molar absorptivity (mol/lit)	4782mol/lit
Recovery studies	
For Tablets	99.80%
For Oral films	99.60%
Precision (%RSD)	
For Tablets	Less than 2 (0.531)
For Oral films	Less than 2 (0.501)

**Table:2 Percent recovery (accuracy) data of Sumatriptan succinate**

Formulation	Formulation Amount ( $\mu\text{g/ml}$ )	Level Of Addition (%)	Amount Added ( $\mu\text{g/ml}$ )	Amount Recovered ( $\mu\text{g/ml}$ )	%Recovery	Average % Recovery
Tablets	105.5	80	84.4	189.4	99.7	99.8%
	105.5	100	105.5	210.6	99.8	
	105.5	120	126.6	231.9	99.9	
Oral films	60	80	48	107.4	99.4	99.6%
	60	100	60	119.6	99.6	
	60	120	72	131.6	99.7	

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